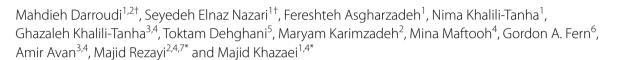
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Fabrication and application of cisplatin-loaded mesoporous magnetic nanobiocomposite: a novel approach to smart cervical cancer chemotherapy



[†]Mahdieh Darroudi and Seyedeh Elnaz Nazari have equal contribution and are co-first authors

*Correspondence: rezaeimj@mums.ac.ir; khazaeim@mums.ac.ir

Metabolic Syndrome Research center, Department of Physiology, Faculty of Medicine, Mashhad University of Medical Science, Mashhad, Iran

² Department of Medical Biotechnology and Nanotechnology, School of Science, Mashhad University of Medical Science, Mashhad, Iran

Full list of author information is available at the end of the article

Abstract

There are significant challenges in developing drug carriers for therapeutic perspective. We have investigated a novel nanocarrier system, based on combining functionalized magnetic nanocomposite with Metal-Organic Frameworks (MOFs). Magnetic nanoparticles modified using biocompatible copolymers may be suitable for delivering hydrophobic drugs, such as cisplatin. Furthermore, compared to polymeric nanocarriers, nanocomposite constructed from zeolitic imidazolate framework-8 (ZIF-8) have demonstrated better drug loading capacity, as well as excellent pH-triggered drug release. Cisplatin-encapsulated Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA has been evaluated to determine the antitumor effects of free cisplatin enhancement in cervical cancer cells. In order to increase the stability of the proposed nanocarrier in aqueous solutions, in addition to the density of functional groups, a nano-chitosan layer was coated on top of the magnetic nanocomposite. It was then added with cisplatin onto the surface of Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA to deliver anticancer treatment that could be targeted using a magnetic field. A mouse isograft model of TC1 cells was used to evaluate the in vivo tumor growth inhibition. In tumor-bearing mice, Fe₃O₄@SiO₃-ZIF-8@N-Chit-FA-cisplatin was injected intraperitoneally, and the targeted delivery was amplified by an external magnet (10 mm by 10 mm, surface field strength 0.4 T) fixed over the tumor site. Based on in vivo results, cisplatin-Loaded Mesoporous Magnetic Nanobiocomposite inhibited the growth of cervical tumors (P < 0.001) through the induction of tumor necrosis (P < 0.05) when compared to cisplatin alone. With the application of an external magnetic field, the drug was demonstrated to be able to induce its effects on specific target areas. In summary, Fe₃O₄ @ SiO₂-ZIF-8 @ N-Chit-FA nanocomposites have the potential to be implemented in targeted nanomedicine to deliver bio-functional molecules.

Keywords: Drug delivery, Cervical cancer, Magnetic nanocomposite, In vivo



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Introduction

Cervical cancer is ranked fourth among most common malignant gynecological tumors. According to the CDC, 311,000 deaths occurred from the 570,000 cases diagnosed in 2018 (Arbyn et al. 2020). Nearly half of all deaths from cervical cancer occurred in women over 58 years of age, and cervical cancer is the second most common death cause in women 20–39 years of age (LaVigne et al. 2017; Siegel et al. 2019). However, late-stage disease and cervical adenocarcinomas remain major concern to women globally and cannot be detected by cytology (Ramezani Farani et al. 2022). Although cytology is the most widely used method for detecting cervical cancer, vaccination against human papillomavirus has proven to be very effective (Adiga et al. 2021). Despite existing trials, cervical cancer remains one of the most common causes of cancer-related deaths in women, and further research is required (Paskeh et al. 2021).

Cisplatin has a central role in cancer chemotherapy; however, its systemic toxicity poses a significant obstacle (Zhao et al. 2015). Cisplatin is used as a chemotherapy agent alone or in combination with other drugs to treat various solid tumors, such as reproductive system tumors, cervical, ovarian, head and neck tumors, and melanomas (Chon et al. 2012; Luo et al. 2021; Maillard et al. 2020; Mitsudomi et al. 2010). Despite the popularity of cisplatin in the cancer treatment, it has serious side effects that include bone marrow suppression, nephrotoxicity, neurotoxicity, dose toxicity, and hearing loss (He et al. 2022; Kuang et al. 2012; Ramkumar et al. 2021). Currently, the formulation of cisplatin is promising although some draw backs, such as short aqueous half-life, ease of oxidation and hydrolysis, and light instability, are problematic. Hence, many nanotechnology approaches, which include green methodologies, microfluidics (Rabiee et al. 2021), microporous intelligent nanostructures (Rahimnejad et al. 2021a), high-gravity approaches, and combinations (Rahimnejad et al. 2021b), are being investigated to determine if they can significantly enhance the drug's effectiveness and efficiency, reduce the cost of its development, increase its eco-friendliness, and speed up the drug discovery process (Jafari et al. 2022). Medical nanotechnology applications have driven the design of novel intelligent nano-systems, which can respond physically and morphologically to the pathological environment of tumor tissues (Gonciar et al. 2019; Huang et al. 2016; Kashyap et al. 2021; Rashid and Zaid Ahmad 2018). This has prompted the exploration of novel nano-formulations in order to overcome the limitations of conventional cancer treatments by addressing disease-specific mechanisms and properties (Ulbrich et al. 2016). The use of drug carriers significantly increases targeted drug accumulation (Tan et al. 2020) in specific organs or cell types and permits controlled release (or activation) of drugs where therapeutic effect is desired, e.g., in the tumor matrix and/or in tumor cells. Furthermore, selective activation could limit toxicity, preventing or minimizing harmful side effects (Pecorino 2021).

In this scenario, researchers have been investigating the properties of magnetic nanoparticles (Abedi et al. 2019; Boyer et al. 2010), which make role due to its various applications in Magnetic Resonance Imaging (MRI) (Smith and Gambhir 2017), magnetic separation (Guo et al. 2017), drug delivery (Darroudi et al. 2021; Vangijzegem et al. 2019), and hyperthermia chemotherapy (Vamvakidis et al. 2018b, 2018a). A magnetic field can be used to manipulate superparamagnetic nanoparticles. When the magnet is removed, there is no residual magnetization, which provides colloidal stability and

dispersibility for the NPs, thereby rendering it suitable for biomedical applications (Lee et al. 2015). The biocompatibility, biodegradability, aqueous dispersibility, and magnetic properties of magnetic NPs make them good candidates for biomedical applications (Zanganeh et al. 2016). Furthermore, the development of nanobiomaterials for drug delivery functions offers new possibilities for individualized diagnosis and treatment (Cheng et al. 2018; Yang et al. 2018; Zou et al. 2018). Metal-organic framework materials (MOFs) have a range of advantageous properties, including flexible modification, tunable pore size, and controllable synthesis (Rojas et al. 2019; Wang 2017). MOFs can also offer multiple versatilities for encapsulating and loading active ingredients (Qin et al. 2019). Mesoporous nanoscale zeolitic imidazolate framework-8 (ZIF-8) is a wellknown MOF versatile nanocarrier due to its biocompatibility and biodegradability upon acidic conditions (Lv et al. 2019; Zheng et al. 2016). Through functionalizing nanocarriers, pharmacokinetics and biodistribution can be modified (Mandriota et al. 2019). In addition, the use of polymers as a vehicle for drug delivery has been receiving considerable attention in recent years due to their designability, biocompatibility, and other properties (Nejadshafiee et al. 2019). Folic acid (FA) is also an ideal targeting molecule since folate receptors are frequently expressed on cancer cells and rarely found on normal cells (Zadeh Mehrizi et al. 2018); the use of folic acid is facilitated by its natural adhesive properties and functional groups, which are able to form thin, surface-adherent polymer films onto virtually all material surfaces (Azizi et al. 2021). Phase separation and ionic gelation methods were mainly implemented to prepare nano-chitosan biocomposites (Ramezani Farani et al. 2022; Vaezifar et al. 2013). Its functionalization with folic acid is considered a good substance for the delivery of molecules due to its high solubility in water, excellent biocompatibility, and biodegradability (Jermy et al. 2021, 2019). Upon further investigation, it was found that nano-chitosan functionalized by FA behaved highly dependently on the pH value of the solution (Farahnak Roudsari et al. 2021). Herein, we present a bioinspired mesoporous and magnetic nanocomposite containing Chitosan NPs-FA-functionalization (Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA) as a targeted drug delivery system (Fig. 1).

Materials and methods

Materials

All chemicals were obtained from commercial supplier and, except otherwise stated, used without further purification. Toluene, sodium tripolyphosphate (TPP), Chitosan (Chit), phosphate buffer saline (PBS), sodium dodecyl sulfate (SDS), high glucose cell culture medium, and a tetrazolium-based assay were acquired from Sigma-Aldrich. PBS at pH 7 was prepared by mixing monobasic potassium phosphate (K_2PO_4) and dibasic potassium phosphate (K_2PO_4) in an aqueous solution. Throughout all experiments, ultrapure water was used as a water system.

Preparation of nano-chitosan

The chitosan NPs containing FA were synthesized by Roudsari et al. with some modifications (Alishahi et al. 2011; Farahnak Roudsari et al. 2021). A transparent chitosan solution was prepared by dissolving chitosan in 1% (w/v) acetic acid using a magnetic stirrer, followed by diluting with deionized water to create a solution with a final concentration

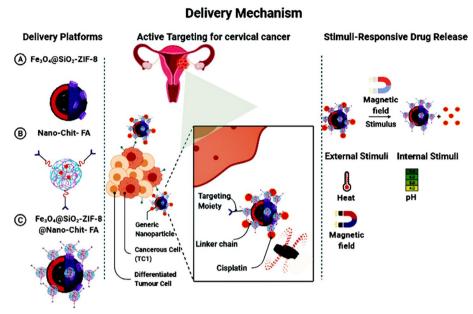


Fig. 1 a Fabrication of Fe_3O_4 @SiO₂-ZIF-8@N-Chit-FA, **b** Schematic of in vivo targeted drug delivery of cisplatin against cervical cancer

of 0.1 (w/v) at pH 5. TPP solution at a concentration of 0.05% (w/v) was added dropwise with an equal volume of chitosan solution for 15 min to induce the spontaneous formation of chitosan—TPP nanoparticles by ionic gelation mechanism. To prepare chitosan—TPP nanoparticles coated with FA, the suspension was stirred by stirring for 60 min at room temperature, followed by crosslinking the amino groups of chitosan—TPP and FA. We isolated the nanoparticles by centrifuging the solution at 8000 rpm for 30 min and drying the precipitate in a freeze dryer.

Synthesis of Fe₃O₄@SiO₂@ZIF-8 nanoparticles

A facile solvothermal method was used to prepare the magnetic Fe_3O_4 @Si O_2 core—shell (Darroudi et al. 2020; Ghasemi et al. 2021). Afterward, a solution of $Zn(NO_3)_2$.6 H_2O and 2-methyl imidazolate in methanol, 15 mg of Fe_3O_4 @Si O_2 core—shell, was used for 1 h at 60 °C, stirring vigorously throughout the reaction. Following cooling to room temperature, the mixture was separated using a magnet and washed several times with methanol, and the precipitate was dried with a freeze dryer.

Cisplatin loading (Fe₃O₄@SiO₂@ZIF-8@N-Chit-FA-cisplatin)

In order to loading of cisplatin, a ratio of Fe $_3$ O $_4$ @SiO2-ZIF-8@N-Chit-FA (100 ug/mL) and 1 mM aqueous cisplatin (10 mg/10 ml) were tested at a time of 48 h according to Mandriota et al.(Mandriota et al. 2019). The Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA-cisplatin complexes were resuspended in 0.1 M of PBS buffer at pH 7 for stabilization, stirring at approximately 1000 rpm at room temperature. The magnetic nanocarrier was washed three times with external magnets and then resuspended in ultrapure water to remove the remaining cisplatin. The following equation is used to determine the loading efficiency of the drug:

Encapsulation efficiency (%) =
$$(C_{\rm en} - C_{\rm in}) \times 100$$

In which C_{en} and C_{in} represent the amount of encapsulated cisplatin over its input quantity during the process (mg).

Cisplatin release from Fe₃O₄@SiO₂@ZIF-8@N-Chit-FA

Cisplatin release from $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA nanocarriers at 37 °C during a 4-h-period under different pH conditions for various amounts of time was measured, then centrifuged to calculate the amount of released cisplatin (Günday et al. 2020; Vandghanooni et al. 2020). Using an absorbance at 290 nm, the amount of released cisplatin was calculated. In order to calculate the drug release ratio for cisplatin, the following equation has been used:

Adsorption (%) =
$$C_0 - C_1/C_0 \times 100$$

In which C_0 and C_1 correspond to the initial concentration of cisplatin (mg) – free cisplatin concentration (mg).

In vivo intraperitoneal administration under an external magnetic field

Fe₃O₄ @ SiO₂-ZIF-8 @ N-Chit-FA-cisplatin was tested on TC1 tumor models to explore its anticancer properties. C57BL/6 mice (20-30 g) were purchased from the Laboratory Animal Center of Mashhad University of Medical Sciences (MUMS), Mashhad, Iran. An Ethical Committee at the Experimental Animal Center of MUMS, Mashhad, Iran, approved all protocols for animal experimentation. During the experiments, the mice were housed in a laboratory environment (with 12 h of light and 12 h of darkness), and standard food and water were provided. The mice were injected subcutaneously with cell suspensions $(1 \times 10^6/100 \,\mu\text{L})$ of TC1 cells suspended in RPMI-1640 medium and approximately two weeks later; when the tumors reached a size of 80-100 mm³ (Ghaemi et al. 2012) they were divided into three groups randomly as follows (n=6 in each group): I. Control (Untreated group), II. Cisplatin (treated with 5 mg/kg twice at a 3-day interval, intraperitoneal; IP), and III. Fe₃O₄ @ SiO₂- ZIF-8 @ N-Chit- FA-cisplatin (treated with 5 mg / kg twice at a 3-day interval, IP), after which a circular external magnet (10 mm by 10 mm, surface field strength 0.4 T) was fixed over the tumor area. Tumor measurements were conducted every other day on the animals. The tumor volume (V) was calculated according to the formula V = AB2/2, where A is the length of the major axis and B is the length of the minor axis. To calculate the tumor volume (V), we used the following formula V=AB2/2 (A=the major axis length and B=the minor axis length (Asgharzadeh et al. 2022). A cervical tumor was collected from each mouse on day 18 for further investigation by hematoxylin and eosin (H&E) staining methods.

Characterizations

We characterized the materials using UV–Vis spectrometers (Perkin Elmer Lambda 25) to measure absorbance from 200 to 800 nm.20,26 FT-IR spectra in the 400–4000 cm⁻¹ range were obtained using a JASCO FT-IR-460 spectrometer. Structured nanocomposites were characterized by field emission scanning electron microscopy (FESEM) (FESEM, TESCAN MIRA-3) operating at 17 kV and transmission electron

microscopy (TEM) using a JEOL at 300 kV (JEM2100F) for analyzing the morphology and size of synthesized nanomaterials. Powder X-ray diffraction (XRD) spectra were collected using Cu Ka radiation at 1.54056 A wavelength using a Philips PW1730 diffractometer. The magnetic properties of magnetic nanocomposites were determined by vibrating sample magnetometers (Kashan University, Kashan, Iran).

Results and discussion

Figure 2(a-d) depicts the XRD patterns of Fe₃O₄@SiO₂, Fe₃O₄@SiO₂-ZIF-8, Fe₃O₄@ SiO₂-ZIF-8@N-Chit-FA, and Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA-cisplatin. The cisplatin loading over the Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA nanocomposite was carried out using an enforced adsorption technique. A characteristic sharp peak was observed between 15 and 30 degrees due to the Metal-organic framework ZIF-8. The Fe₃O₄ presence was expected to be observed at 2 theta value of 35.7° in all the nanocomposite structures. The presence of weak XRD peaks was observed corresponding to the core-shell structure of Fe₃O₄ (Fig. 2a, b). Such weak peaks are due to the presence of small nanosized Fe₃O₄, which indicates that nanoporous space was not crystallized inside cages of nanocarriers. The metal-organic framework ZIF-8 has a cage-like structure with many pores, depicting the broad peak in the spectra. However, still, the peak intensities at 35.5°, 43.3°, 53.5°, and 62.8° increased over Fe₃O₄-SiO₂ corresponding to (310), (400), (422), and (440) °, respectively (Fig. 2c). Fe₃O₄ particles tend to have different surface modifications depending on the structure of the silica and MOF structure. Diffraction patterns of the Pt complex were observed in cisplatin (Fig. 2d). A broad diffraction peak was associated with Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA-cisplatin, indicating MOF and silica in the amorphous phase. There are no cisplatin peaks in the diffraction patterns, indicating a transformation of cisplatin due to confinement in the cubic pores of silica and MOF, enhancing cisplatin solubility.

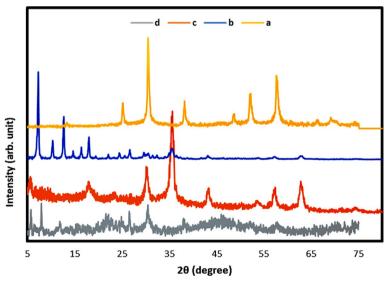


Fig. 2 XRD analysis of **a** Fe₃O₄@SiO₂, **b** Fe₃O₄@SiO₂-ZIF-8, **c** Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA, and **d** Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA and

Through FESEM, the surface morphologies of parent and $Fe_3O_4@SiO_2$ -ZIF-8, $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA, and $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin, and cisplatin alone were examined (Fig. 3a–e). The parent $Fe_3O_4@SiO_2$ was found to be composed of nanosized, spherical spheres of about 27 nm. As a result of the high surface area, iron oxide loaded over SiO_2 did not lead to significant changes in the morphological characteristics compared to the parent material (Fig. 3a). In comparison,

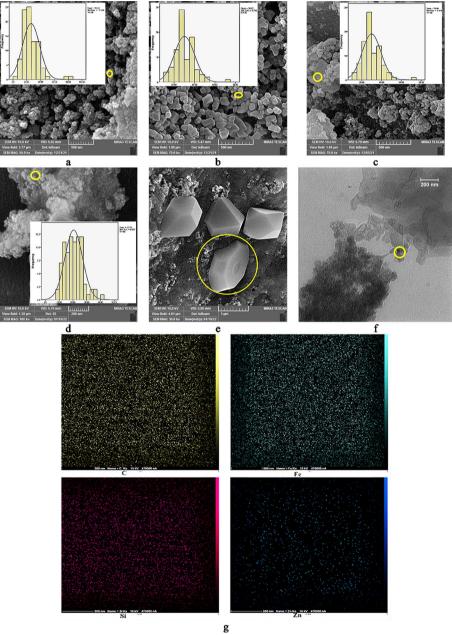
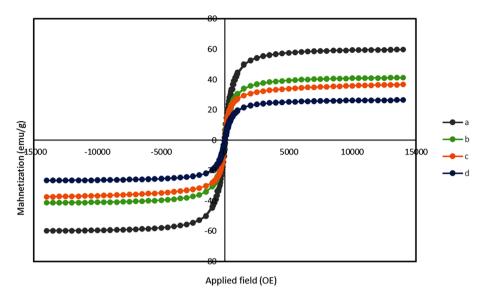


Fig. 3 FESEM analysis of **a** Fe $_3$ O $_4$ @SiO $_2$, **b** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8, **c** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA, and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA-cisplatin, **e** cisplatin alone (inset: frequencies of particle size and average particle diameter histogram), **f** TEM analysis of Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA, and **g** Map of Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA (C, Fe, Si, and Zn)

the Fe₃O₄@SiO₂-ZIF-8 displayed monodispersed MOF distributed uniformly in the range of about 43 nm. The presence of a hetero nanosized framework over the Fe₃O₄@SiO₂ core–shell was observed for this combination of Fe₃O₄@SiO₂-ZIF-8 (Fig. 3b). In the case of Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA nanocomposite and encapsulated cisplatin on the surface of the nanocomposite, the sample had some porous morphological changes with nanosphere structures at a scale bar of 65, and 55 nm, respectively (Fig. 3c, d). Figure 3f depicts the hexagonal morphology of cisplatin alone used in the *in vivo* studies. Compared with cisplatin alone and nanocomposites, it was demonstrated that cisplatin using *in vivo* studies has shown that nanocomposites have unvarying porous morphology with sizes between 21 and 37 nm, while cisplatin alone has the particle diameter out of nanometer size (around 1500 nm).

According to the TEM analysis at magnifications of 45 nm, Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA nanocomposites have unique deposition characteristics and dependency on the support with the dispersion on the porous core-shell structure of the nanocarriers (Fig. 3f). The particles were grouped based on the particle size: 0-20, 21-30, 31-40, 41-50, 51-60, 61-70, and 71-80 nm. The particle size distribution of Fe₃O₄@SiO₂, Fe₃O₄@SiO₂-ZIF-8, Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA, and Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA-cisplatin was as follows: Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA > Fe₃O₄@ SiO_2 -ZIF-8@N-Chit-FA-cisplatin > Fe_3O_4 @Si O_2 -ZIF-8 > Fe_3O_4 @Si O_2 . images, a large particle size of about 42.0 ± 1.07 nm was associated with Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA, and SiO₂ shells and ZIF-8 cages support about 18 ± 2.03 nm. From the EDAX mapping analysis, the elemental distribution of the Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA nanocomposite is shown in Fig. 3g. The Fe, Si, Zn, and C atoms were found and are shown in different colors. It was demonstrated that the entire atoms had been distributed in the whole nanocomposite structures uniformity (Fig. 3g). This confirms that a part of the Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA clearly shows the distribution of elements and also an immense perspective to be used in the progress of novel magnetic nanobiocomposite.

Vibrating sample magnetometers (VSMs) were used to measure the magnetic properties of nanoparticle composites. Figure 4(a-d) shows the VSM for the systems, respectively: (a) Fe₃O₄@SiO₂, (b) Fe₃O₄@SiO₂-ZIF-8, (c) Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA, and (d) Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA-cisplatin. The magnetization over three different nanocomposites varied in magnitude. According to Fe₃O₄ magnetization generated over Fe₃O₄@SiO₂, Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA, and Fe₃O₄@ SiO₂-ZIF-8@N-Chit-FA-cisplatin nanocomposites, Fe₃O₄@SiO₂ generated the highest value (59.9 emug⁻¹), followed by $Fe_3O_4@SiO_2$ -ZIF-8 (40.3 emug⁻¹), $Fe_3O_4@$ SiO₂-ZIF-8@N-Chit-FA (36.5 emug⁻¹), and Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA-cisplatin (26.3 emug⁻¹). Nanoclusters of smaller sizes have been discovered to form superparamagnetic interactions between iron oxide species in core-shell structure SiO₂, while larger nanoclusters contribute to magnetic behavior. An impregnation of Fe₃O₄ into MOF with large pores produces small nanosized Fe₃O₄ nanocomposites, as is shown in the present study. Based on the TEM (Fig. 3d) study, Fe₃O₄@SiO₂ core-shell was 13 nm in size on average over ZIF-8. The VSM spectrum of Fe₃O₄@SiO₂-ZIF-8 showed superparamagnetic behavior with narrow hysteresis in agreement with TEM analysis. A Fe₃O₄@SiO₂-ZIF-8@N-Chit-FA pore produces medium nanoclusters with



 $\textbf{Fig. 4} \ \ VSMs \ analysis \ of \ \textbf{a} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8, \ \textbf{c} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 - ZIF-8 @ N-Chit-FA, \ and \ \textbf{d} \ \ \text{Fe}_3O_4 @ SiO_2 -$

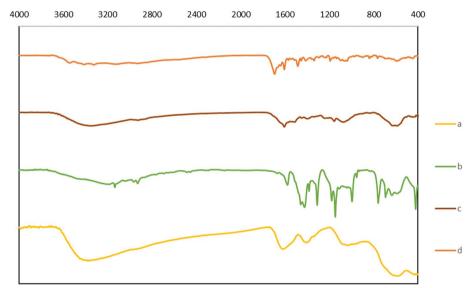


Fig. 5 FT-IR spectra of **a** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8, **c** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA, and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$ @SiO $_2$ -ZIF-8@N-Chit-FA and **d** Fe $_3$ O $_4$

an average particle size of 18 nm (Fig. 3e). In contrast to $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin, $Fe_3O_4@SiO_2$ loaded over monodisperse ZIF-8 produced the strongest magnetization, indicating superparamagnetic behavior.

Following functionalization of magnetic nanoparticle Fe_3O_4 with shells of silane and ZIF-8, FT-IR spectroscopy was used to exhibit interactions and complexes in Fe_3O_4 @ SiO_2 -ZIF-8@N-Chit-FA and Fe_3O_4 @ SiO_2 -ZIF-8@N-Chit-FA-cisplatin (Fig. 5a–d). The spectra analysis of cisplatin revealed strong amine stretching bands at 3338 cm⁻¹ and 3267 cm⁻¹ and a broad peak at 1654 cm⁻¹ as the characteristic peak of cisplatin (Fig. 5). FIT-IR profiles are shown in Fig. 5 for nanocarrier Fe_3O_4 @ SiO_2 , Fe_3O_4 @ SiO_2 -ZIF-8,

 ${\rm Fe_3O_4@SiO_2\text{-}ZIF\text{-}8@N\text{-}Chit\text{-}FA}$, and ${\rm Fe_3O_4@SiO_2\text{-}ZIF\text{-}8@N\text{-}Chit\text{-}FA\text{-}cisplatin}$ following iron oxide and silane core—shells. ${\rm Fe_3O_4@SiO_2\text{-}ZIF\text{-}8@N\text{-}Chit\text{-}FA\text{-}cisplatin}$ had weak amine bands indicating a conjugated NH₂ of nano-chitosan and OH group of nanocomposite and folic acid toward cisplatin. Furthermore, the bending vibration at 1650 cm⁻¹ indicates that cisplatin has been functionalized and the reaction of the folic acid and nano-chitosan and Pt(II) complex was performed.

By utilizing Energy-dispersive, X-ray spectroscopy estimates the percentage of different elements in $\text{Fe}_3\text{O}_4@\text{SiO}_2(a)$, $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-ZIF-8}$ (b), $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-ZIF-8}@\text{N-Chit-FA}$ (c), and $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-ZIF-8}@\text{N-Chit-FA-cisplatin}$ (d). The results have shown that FA functionalized chitosan nanocomposite consists of three main organic elements of N, and O with a weight of 12%, and 31%, and main elements of Fi, Si, and Zn with a weight of 10%, 0.5%, and 2% in the core–shell structure. After the encapsulating of drug cisplatin, atomic percentage of O, Fe, Zn, and Si had changed to 27%, 14, 1.30, and 0.5%, respectively. Furthermore, encapsulating of cisplatin on the surface of $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-ZIF-8@N-Chit-FA}$ nanocomposite was demonstrated through 2% element percentage (See Fig. 6).

Antitumor and pro-necrotic effects of magnetic nanobiocomposite in vivo

A murine tumor model was used to determine if $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin inhibits the growth of cervical tumors. The mice were treated with $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin or cisplatin alone, and tumor weight and size were assessed on each group. According to data from experiments, $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin decreased tumor size and weight in mice (Fig. 7 A, B). $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin exerts an enhanced anticancer activity when compared with cisplatin

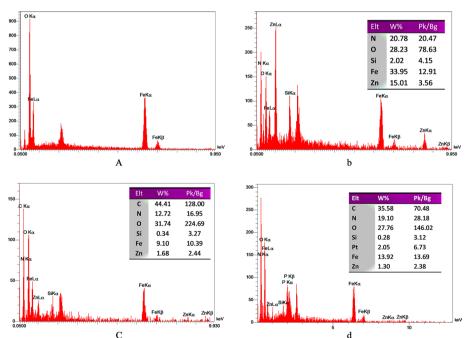


Fig. 6 EDX analysis of particle constituents of **a** Fe_3O_4 @ SiO_2 , **b** Fe_3O_4 @ SiO_2 -ZIF-8, **c** Fe_3O_4 @ SiO_2 -ZIF-8@N-Chit-FA, and **d** Fe_3O_4 @ SiO_2 -ZIF-8@N-Chit-FA-cisplatin

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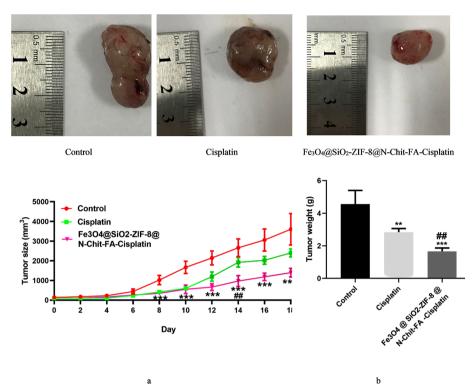


Fig. 7 Fe₃O₄ @SiO₂-ZIF-8 @ N-Chit-FA-cisplatin reducing tumor size and tumor weight in cervical cancer tumor. Effect of Fe₃O₄@SiO₂-ZIF-8 @ N-Chit-FA-cisplatin and cisplatin on tumor size **a** during experiment and tumor weight **b** at the last of experiment in cervical cancer mouse model

alone; consequently, $\mathrm{Fe_3O_4@SiO_2}$ -ZIF-8@N-Chit-FA-cisplatin enhanced cisplatin's antitumor activities in this cervical cancer model. Cisplatin alone follows a one-compartment model in vivo based on the in vivo release profile of nanobiocomposite. As a result, the magnetic nanobiocomposites are composed of multiple compartments that release in bursts and then release continuously.

A challenge in treating cervical cancer is how to prevent the accumulation of cancer-fighting drugs in healthy tissue while improving the local accumulation of anticancer drugs in the tumor sites (Koning et al. 2010). Nanoparticles with magnetic properties may offer new possibilities for localized cancer treatment (Kumar and Mohammad 2011). A polymeric nanocapsule containing 5-Fu has recently been reported to exhibit similar behavior in the treatment of colon cancer (Li et al. 2008). Shakeri-Zadeh et al. found that 5-Fu had increased colon tumor tissue tendency when loaded into magnetic nanoparticles (Shakeri-Zadeh et al. 2014). In this regard, our in vivo experiments have shown that when carbon nanoparticles are loaded with cisplatin, it has a sustained release, prolonged half-life, and increases tumor tissue uptake.

Figure 8A demonstrates that $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin resulted in an increased area of tissue necrosis in the cervix tumor. A comparison of $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin and cisplatin alone as the standard chemotherapeutic regimen in cervical cancer revealed that $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA-cisplatin increased the percent of necrosis in cervical tissues (Fig. 8, B; p0.05). The necrosis area in H & E-stained section is indicated with arrows.

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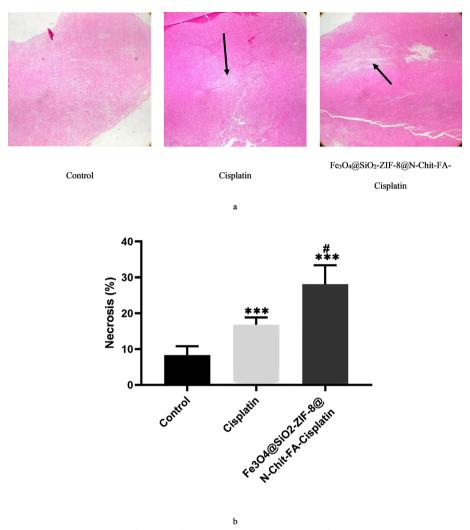


Fig. 8 Fe₃O₄ @SiO₂-ZIF-8@ N-Chit-FA-cisplatin promoting necrosis in cervical cancer tumor. Necrosis areas within tumor shown by H&E staining observed with light microscope (**a**). The arrows indicate necrosis area. Quantification of necrotic area with image J software (**b**)

Conclusion

In summary, we have prepared versatile magnetic nanocarrier $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA, further modifying the structure with targeting ligands has led to promote the targeted delivery of cisplatin to cancerous cells in the biological media. The functionalized and magnetic $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA nanocarrier with average diameter 42.0 ± 1.07 nm exhibited high release pores with large windows. By using external magnetic fields in chemotherapy, $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA nanocomposites were able to prevent drug leakage and enhance cancer treatment efficacy in tumor-bearing mice. Additionally, the magnetic nanocomposite demonstrated excellent antitumor activity against cervical cancer isografts. It may be possible to significantly reduce the adverse effects of cisplatin therapy by delivering it with magnetic nanocomposite and to improve the therapeutic index of cisplatin through our method of delivering cisplatin loaded with magnetic nanocomposite. Overall, using $Fe_3O_4@SiO_2$ -ZIF-8@N-Chit-FA not only

expands the applications of functionalized magnetic MOF nanocomposites but also offers new opportunities for developing effective and safe targeted therapeutic carriers that may help gain insight into cancer treatment.

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Author contributions

MD contributed to visualization, investigation, formal analysis, analysis of data, writing original draft, and methodology. SEN contributed to conceptualization, methodology, visualization, investigation, and formal analysis. FA was involved in analysis of data, methodology, investigation, and writing—original draft. NK-T and TD were involved in methodology and visualization. GK-T and MM contributed to methodology. MK contributed to methodology and writing—review & editing. GAF and AA were involved in writing—review & editing. MR contributed to supervision, funding acquisition, conceptualization, and writing—review and editing. MK was involved in supervision, funding acquisition, conceptualization, writing—review and editing, and validation. All authors read and approved the final manuscript.

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Availability of data and materials

All the data sharing not applicable to this article as no dataset was generated or analyzed during the current study. This manuscript contains all the required data to support the findings of this study.

Declarations

Ethics approval and consent to participate

It was not applicable.

Consent of publication

All the authors contributed to the article and approved the submitted version.

Competing interests

The authors declare no competing interests.

Author details

¹ Metabolic Syndrome Research center, Department of Physiology, Faculty of Medicine, Mashhad University of Medical Science, Mashhad, Iran. ² Department of Medical Biotechnology and Nanotechnology, School of Science, Mashhad University of Medical Science, Mashhad, Iran. ³ Medical Genetic Research Centre, Mashhad University of Medical Science, Mashhad, Iran. ⁴ Metabolic Syndrome Research Centre, Mashhad University of Medical Science, Mashhad, Iran. ⁵ Department of Microbiology and Virology, Faculty of Medicine, Mashhad University of Medical Sciences, Mashhad, Iran. ⁶ Brighton & Sussex Medical School, Division of Medical Education, Falmer, Brighton, Sussex, UK. ⁷ Medical Toxicology Research Centre, Mashhad University of Medical Science, Mashhad, Iran.

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