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# Comparative effect of thermo/pH-responsive polymer-coated gold nanocages and hollow nanostars on chemo-photothermal therapy of breast cancer cells

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# **Abstract**

**Background:** Combination chemo-photothermal therapy appears to be one of the next generations of cancer treatment. In this study hollow gold nanostars (HGNSs) and gold nanocages (GNCs) were synthesized and stabilized with thermo-pH-sensitive thiol-end capped ABC triblock copolymer poly(acrylic acid)-b-poly(*N* isopropylacrylamide)-b-poly (e-caprolactone)-SH; PAA-b-PNIPAAm-b-PCL-SH (GNSs@Pol). Doxorubicin (Dox) was conjugated to the GNSs@Pol nanostructures via ionic interaction, covalent attachment and hydrogen bonding (GNSs@Dox-Pol). The physicochemical characteristics of prepared GNSs@Pol and GNSs were assessed using dynamic light scattering (DLS), transmission electron microscopy (TEM) and zeta potential techniques. Cytocompatibility of the GNSs@Pol was studied by hemolysis assay and MTT assay. The chemo-photothermal therapy (PTT) potential of GNSs@Dox-Pol was compared on MCF7 cells using MTT assay, cell cycle, DAPI staining and Annexin-V apoptosis assay techniques.

**Results:** Cell internalization results showed an almost complete uptake of GNSs@Pol by MCF-7 cells in the first 3 h of treatment. The heat generation measurement results showed that both of GNSs have a potential for light to heat conversion ( $\Delta T = 23-27$  °C) and HGNSs demonstrated better efficiency than GNCs after 10-min exposure to NIR irradiation. Following chemo-photothermal treatment, the highest cell mortality (90%) and apoptotic effects (97% apoptosis) were observed in HGNSs@Dox-Pol received laser irradiation treatment group.

**Conclusions:** This work highlights the potential application of designed GNSs@Dox-Pol in a combinational chemo-PTT to treat breast cancer cells.

**Keywords:** Photothermal therapy, Anisotropic gold nanostructure, Plasmon surface resonance, Combination therapy



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# Introduction

In the clinic, major therapeutic candidates for cancer therapy include surgical resection, immunotherapy, chemotherapy, radiotherapy, gene therapy, and thermotherapy. However, in many cases, the mentioned treatment regimens are not effective and have many associated side effects (Dianat-Moghadam 2018, 2020). For this purpose smart nanoparticles have been generally used as vehicles for delivering anti-proliferative drugs to target cancer cells (Chaudhuri 2009; Chekhun 2009). Previous studies have shown that the mentioned methods are generally deemed less effective in comparison with the combination therapy approach (Jaffray 2012; Li 2013). Besides, combination therapy has a synergistic effect and reduced side effects in cancer therapy. Therefore, in order to improve treatment outcomes, the use of multi-modality is an essential step in cancer therapy (Yap et al. 2013; Blagosklonny 2004). Compared to separate photothermal therapy (PTT) and chemotherapy, combined chemo-photothermal therapy demonstrates significantly improved efficacy in treatment. The reason is more precise control of drug-release patterns and enhanced cytotoxicity of chemotherapy drugs at higher temperatures. A prevalent restriction in regular chemotherapy is a decrease in therapeutic efficacy due to the drug resistance phenomenon (Hauck 2008; Dykman and Khlebtsov 2012), which can be compensated by PTT.

Among different types of nanoparticles, gold nanostructures (GNSs) offer multipurpose structures for the creation of smart nanoparticles. GNSs are good candidates for combination therapy since they can deliver chemotherapy and create cytotoxic heat by PTT for the targeted cancerous cells. GNSs biocompatibility, tunable localized surface plasmon resonance (LSPR), and active surface to functionalization, make them potent nanocarriers for the targeted delivery of numerous anticancer drugs. Moreover, GNSs exhibit fine-tunable LSPR, providing a critical role in optimizing photothermal conversion (Broek 2011). Between a diverse set of GNSs, hollow interior structures, such as hollow gold nanostars (HGNSs) and gold nanocages (GNCs), have been of particular interest for chemo-PPT due to their high surface-to-volume ratio, strong light-heat conversion features, and PTT efficiency due to having multiple sharp spikes as well as high capabilities for drug loading (Melancon et al. 2011; Chen 2007; Zheng 2016; Hao 2007; Halas 2010; Yavuz 2009). The hollow and porous structure of these GNSs can lead to drug leakage into the target sites from early stages (Wang 2014). Besides the mentioned advantages, HGNSs and GNCs have critical challenges such as poor stability and short blood circulation time to use as a cancer therapeutic agent (Mura et al. 2013; Yang 2013). To overcome these problems, modification of HGNSs and GNCs using polymer hybrid systems is one of the suitable strategies.

The design of conjugated HGNSs and GNCs with amphiphilic copolymers can lead to an increase in their stability. This can also provide a unique and tailorable platform for the targeted delivery of chemotherapeutic drugs in combination with PTT. In the stimuli-responsive polymeric systems, the cargo can be released by physical (light, sound, pressure, temperature, electricity, or magnetic field) or chemical stimulations (pH, ions composition or activation by specific molecules) of the environment (Ding and Li 2017; Xiang 2017). GNSs with generated hyperthermia through NIR laser irradiation accelerated the release of anticancer drugs into the target sites (Halas 2010; Gao 2012).

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In the present work, we aimed to engineer a novel combination system that benefits from the therapeutic efficacy of HGNSs and GNCs as photothermal transducers and doxorubicin (Dox) as an antitumor drug. At first, GNSs were synthesized and stabilized by covering their surface with a dual thermo-pH-sensitive micelle system (GNSs@Pol) composed of poly (acrylic acid)-b-poly (*N*-isopropyl acrylamide)-b-poly(e-caprolactone)-SH; (PAA-b-PNIPAAm-b-PCL-SH)] (Abbasian 2017). Then Dox was conjugated to GNSs@Pol via ionic interaction hydrogen and amid bonding. Dox release from the GNSs@Dox-Pol micelle was accelerated under mild acidic conditions (tumor cells endosome) and excitation with near-infrared (NIR) laser. In the second step, the chemo-photothermal therapy efficacy of HGNSs@Dox-Pol and GNCs@Dox-Pol PTT in MCF-7 breast cancer cells was compared.

## **Materials and methods**

#### **Materials**

Sodium borohydride (NaBH<sub>4</sub>), L-ascorbic acid (AA), and silver nitrate (AgNO<sub>3</sub>) were purchased from local chemical companies. Cetyl trimethylammonium bromide (CTAB, 99%), poly(vinyl pyrrolidone) (PVP, powder, average Mw  $\approx$  40,000) and HAuCl<sub>4</sub>.3H<sub>2</sub>O (99.9%) were bought from Sigma-Aldrich Co. (St. Louis, MO, USA). Ethylene glycol, Na<sub>2</sub>S.9H<sub>2</sub>O, acetone, ethanol (reagent grade) were purchased from Merck. Doxorubicin hydrochloride was purchased from Zhejiang Hisun Pharmaceutical Co., Ltd., Taizhou, China. 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), 4′,6-diamidino-2-phenylindole (DAPI), fetal bovine serum (FBS), Roswell Park Memorial Institute 1640 growth medium (RPMI), propidium iodide (PI), rhodamine B, and ribonuclease A were obtained from Gibco BRL Life. Millipore Synergy ultrapure water (DI) of resistivity = 18.2 M $\Omega$  cm<sup>-1</sup> was used for the synthesis of HGNSs and GNCs.

#### Instrumentation

Absorption spectra were recorded using double beam PC 1650 UV–Vis (SHIMADZU, Kyoto, Japan) spectrometer. Morphology and particle size of gold nanostructures were analyzed by transmission electron microscopy (TEM, LEO 906 at 100 kV acceleration voltages, Zeiss, Germany). To obtain average nanoparticle size and zeta potential, dynamic laser scattering was used (Zetasizer Nano, Malvern, UK). Inductively coupled plasma mass spectrometry (ICP-MS; Elan 6100DRC-e, Perkin-Elmer, Waltham, MA, USA) was used for determination of Au ion concentration and quantification of GNSs internalization by MCF-7 cells.

# Preparation of gold nanostructures

# Synthesis of ABC triblock copolymer

Thiol-end capped ABC triblock copolymer poly(acrylic acid)-b-poly(N isopropylacrylamide)-b-poly (e-caprolactone)-SH; PAA-b-PNIPAAm-b-PCL-SH was synthesized in four steps with combination of ring-opening polymerization and RAFT polymerization. At first bis(2-hydroxyethyl) disulfide was synthesized with dimerization of mercaptoethanol. Then The (PCL-S)2 was synthesized by ring-opening polymerization (ROP) of  $\varepsilon$ -caprolactone monomer using bis(2-hydroxyethyl)disulfide as an initiator. Due to its applicability to all forms of vinyl monomers the RAFT procedure used for the

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synthesis of ABC triblock copolymers. Therefore (PAA-b-PNIPAAm-b-PCL-S)2 pent-ablock copolymer was synthesized through the RAFT technique and finally thiol-end capped copolymer (PAA-b-PNIPAAm-b-PCL-SH) was obtained by breaking S–S bound of copolymer with dithiothreitol (DTT) method. The synthesis protocol of all four step was explained in detailed in previously published articles (Abbasian 2017; Mahmoodzadeh 2017).

# Synthesis of gold nanocages (GNCs)

Au–Ag alloyed nanocages were synthesized in the two-step procedure described previously (Skrabalak 2007). First, Ag nanocubes were prepared by addition of poly(vinyl pyrrolidone) (PVP, 10 ml, 300 mg/ml in EG, MW  $\approx$  40 000) to EG in an oil bath as precursor solution. Then AgNO $_3$  (2.5 mL, 282 mM in EG) slowly was injected into the heated solution. The obtained Ag nanocubes were converted to GNCs via a galvanic replacement reaction between silver and gold nanoparticles (Wang 2014; Khlebtsov 2013). An aqueous solution of HAuCl $_4$  (0.2 mM, 10 ml) was titrated via a syringe pump at a rate of 0.8 mL/min into 0.5 mL of synthesized silver nanocube until the color of the reaction was stable. Because of the reaction stoichiometry for the galvanic replacement reaction (i.e., oxidation of three Ag atoms produces one gold atom), hollow structures are produced.

## Synthesis of hollow gold nanostars (HGNSs)

The hollow gold nanostars (HGNSs) were synthesized via a seed-mediated growth method (Wang 2012). First, silver seed nanoparticles were synthesized as sacrificial templates which provides rough surface and active sites for the further deposition of gold nanoparticles. Briefly  $AgNO_3$  was dissolved in 25 mL of deionized water and preheated at 80 °C. Then sodium citrate (1% in mass, 1.5 mL) added to initiate the reduction of  $AgNO_3$  and the temperature was increased to 110 °C. Next, for synthesis of HGNSs,  $HAuCl_4$  aqueous solution (1 mL, 2.94 mM) and AA (10 mM, 1 ml) slowly injected. The procedure was complete by galvanic replacement reaction between silver nanoparticles and  $HAuCl_4$ .

# Coating GNSs with ABC triblock copolymer (GNSs@Pol)

For preparation of both HGNSs and GNCs-coated polymers, each PVP-covered GNSs was separately dissolved in 2 mL of DI water. Then, a 10-mL aqueous solution of PAA-b-PNIPAAm-b-PCL-SH copolymer (0.20 g) to each of them and the mixture was shaken at 600 rpm for 5 days. PVP was exchanged by PAA-b-PNIPAAm-b-PCL-SH copolymer. The solution was then centrifuged (11,000 rpm, 4 °C, 10 min) and the supernatant was discarded. Finally, the polymer-covered GNSs were washed with DI water (Yavuz 2009).

# Photothermal heating setup

The photothermal effect of the GNSs was evaluated by observing temperature increase as a function of time. After exposing the GNSs to NIR continuous laser (808 nm wavelength diode laser, 2.5 W, model: PSU-III-LED, Changchun new institute, China), the temperature rise was recorded. For evaluation of PTT, aqueous solutions of GNSs@Pol were diluted to a desired final concentration (100 µg mL<sup>-1</sup>) in phosphate-buffered saline

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(PBS; Sigma-Aldrich). Then, a certain amount of different aqueous suspension (100  $\mu$ l) was placed in a single well of a 96-well plate and irradiated with 808 nm continuous-wave NIR laser at a power density of 0.7 W/cm² for exposure times of 2, 4, 6, 8 and 10 min. The temperature of the suspensions was evaluated during irradiation by a digital thermocouple (LCD K-Type Digital Thermometer, Model: TM-902C w Thermocouple Wire, ShenZhen, China) immersed into suspensions.

# Dox-loading and release studies

# Dox-loading on GNSs@Pol

To load Dox into GNSs,  $10 \mu L$  aqueous solutions of Dox.HCl ( $2000 \mu g \ mL^{-1}$ ) were added directly to the polymer-covered GNSs (GNSs@Pol;  $1 \ mL$ ,  $200 \ \mu g \ L^{-1}$ ). The mixture was stirred at room temperature for about  $48 \ h$  under the dark situation. The unloaded Dox was removed by centrifugation ( $11,000 \ rpm$ ,  $4 \ ^{\circ}C$ ,  $10 \ min$ ) and washing with DI water twice. The obtained product was denoted as GNSs@Dox-Pol. The loading efficiency of Dox was quantified using UV–vis spectroscopy (Jang 2014):

$$Doxen cap sulation efficiency \left(\% \frac{w}{w}\right) = \frac{amount of Doxinn an oparticle}{amount of feeded Dox}. \tag{1}$$

#### Effect of pH on the Dox release from GNSs@Dox-Pol

In order to evaluate pH-responsive Dox release from GNSs@Dox-Pol, 1 mL of each GNSs@ Dox-Pol suspensions were dispersed in 1 mL of PBS solution with three different pH values (4, 5.4, and 7.4) and shaken in an incubator at 37 °C for definite times. At distinct time intervals, GNSs@ Dox-Pol were collected by centrifugation (11,000 rpm, for 10 min) for recognition of released Dox. After that, equal amount of fresh medium of PBS solutions were added to resuspend the GNSs@ Dox-Pol microtube. Released drugs were measured from the supernatant at 480 nm using an ultraviolet–visible spectral measurement by the following formulae:

$$Drugrelease(\%100) = \frac{amount of Dox release in a medium}{amount of Dox loaded in nanoparticles} \times 100. \tag{2}$$

# Effect of laser irradiation on the release of Dox from GNSs@ Dox-Pol

For each data point, 1 mL of each GNSs@Dox-Pol transferred to a 1-mL centrifuge tube. Before laser irradiation, the samples were shaken and preheated in an incubator to 37 °C. Then each sample was separately irradiated by a 808 nm laser for 10 min (0.7 W/cm²) at defined time intervals. After exposure, samples were collected by centrifugation (11,000 rpm for 10 min). The percentage of released Dox in the supernatant was determined by an ultraviolet–visible spectral measurement at 480 nm (Eq. 2).

# Hemolytic activity test

To evaluate, the biocompatibility of GNSs@Pol, hemolytic activity test (red blood cells, RBCs) was performed according to the recent literature (Zhou 2015). For this purpose, human fresh blood samples were (taken from Iranian Blood Transfusion Institute (IBTI)) diluted with PBS and centrifuged (2500 rpm for 8 min, using an EDTA tube for

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stabilized RBCs) (Rahimi 2017). After removing the supernatant, RBCs were collected and washed several times with PBS (pH 7.4). The dilution of collected RBCs were then performed with PBS, from which 0.5 mL was added to 0.5 mL PBS (as negative control), and to 0.5 mL DI water (as positive control). The same amount of diluted RBCs suspension was then added to GNSs@Pol suspension with different concentrations (1.5, 3, 6.5, 12.5, 25, 50 and 100  $\mu$ g mL<sup>-1</sup>). After 3 h of incubation and shaking (at 37 °C), samples were investigated under an optical microscope using a magnification of × 100 for hemocompatibility. Finally, samples were centrifuged (5000 rpm for 5 min) and released hemoglobin absorption at 540 nm was then analyzed for the supernatants (the experiment was repeated two more times) (Rahimi et al. 2017).

Hemolysis rate of samples were measured using the following formula:

$$Hemolysis(\%) = \frac{(Abs - Ab0)}{(Abs100 - Abs0)} \times 100\%, \tag{3}$$

where Abs, Abs0, and Abs100 are the supernatant absorbance of samples, negative, and positive control, respectively.

# Cellular uptake studies

## Cellular uptake assessment by flow cytometry and fluorescence microscope

For quantitative evaluation of cellular uptake of bare GNSs and GNSs@Dox-Pol nanostructures, we have used flow cytometry method according to a previous report (Poursalehi 2019). For this purpose, rhodamine B-loaded GNSs and GNSs@Dox-Pol nanostructures were treated with MCF-7 cells. After 1 and 3 h, the cells were trypsinized and washed with PBS and counted with a FACScalibur flow cytometer (Becton Dickinson Immunocytometry Systems, San Jose, CA, USA) to measure fluorescent intensity. For further confirmation of cellular uptake of samples, fluorescence microscope (Olympus microscope Bh2-RFCA, Japan) images were captured for qualitative assessment.

## In vitro cell uptake assay by ICP-MS

The intracellular content of GNSs was further evaluated using inductively coupled plasma mass spectrometry (ICP-MS) (Dykman and Khlebtsov 2012). Approximately 100,000 cells were seeded into a 12-well plate in complete RPMI 1640 media and incubated at 37 °C for 24 h. After reaching ~80% confluency, MCF-7cells were treated with samples (GNCs, GNCs@Dox-Pol, HGNSs, and HGNSs@Dox-Pol) in FBS-free RPMI 1640 for 3 h. After 3 h, the cells were extensively washed with ice-cold PBS four times to remove any nanoparticles that were still bound to the cell surface. Studies showed that four wash cycles were sufficient to remove essentially all GNSs bound to the cell surface, leaving only the internalized GNSs (Hou 2016). MCF-7 cells were trypsinized and washed with freshly aqua regia (1 mL) to completely digest the cells and dissolve the gold nanostructures. The resulting solution was incubated at room temperature for 24 h and then heated to remove aqua regia. To determine gold content by ICP-MS (Perkin-Elmer, Waltham, MA, USA) residual was dissolved in DI water.

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# Cell viability assay

#### Cell viability assay before photothermal treatment

For in vitro cytotoxicity studies, breast adenocarcinoma (MCF-7) cells were seeded in 96-well plates (1 × 104 cells/well) in RPMI 1640 supplemented with 10% FBS and 1% benzylpenicillin/streptomycin and incubated at 37 °C in 5% CO<sub>2</sub> overnight. Dox, HGNSs, HGNSs@Pol, HGNSs@Dox-Pol, GNCs, GNCs@Pol, and GNCs@Dox-Pol with different concentrations (1.5, 3, 6.5, 12.5, 25, 50 and 100  $\mu$ g mL<sup>-1</sup>) were diluted with complete RPMI 1640 medium and added to MCF-7 cells and further incubated for 48 h. After that cell media were removed carefully and MTT solution in cell growth culture (200  $\mu$ L, 5  $\mu$ g mL<sup>-1</sup>) was added to each well and incubated for 4 h at 37 °C. Cells without treatment were regarded as negative controls. The optical density of the medium was measured using an ELISA plate reader (Awareness Technology, Palm City, FL, USA) at  $\lambda$  = 595 nm.

# Cell viability assay after photothermal treatment

To estimate the effectiveness of PTT, MCF-7 cells were treated with  $100~\mu g~mL^{-1}$  of GNSs@Pol, and the GNSs@Dox-Pol nanostructures. After 4 h, the medium was removed carefully and washed with PBS to eliminate un-uptaken GNSs. Culture medium was added again and the cells were irradiated (with NIR laser, 808 nm, 0.7 W for 10 min; which from now on will be called irradiation for brevity). The cells that were exposed to only laser irradiation were considered as positive controls. Cells without treatment were regarded as negative controls. Cell viability was assessed by MTT assay method as described above.

# Cell cycling

The effect of GNSs on the cell cycle in different phases is determined with reagents including DNA dyes such as propidium iodide (PI). MCF-7 cells were seeded in the 6-well plate ( $50 \times 10^4$  cells/well) and incubated for 24 h. Then, the cells were treated with the IC<sub>50</sub> concentration of samples (GNCs, GNCs@Dox-Pol, HGNSs, HGNSs@Dox-Pol, and free Dox). After 4 h, the medium was removed carefully and washed with PBS to eliminate un-uptaken GNSs. Culture medium was added again and incubated for 24 h. Then, cells were irradiated and were further incubated for 3 h. After that, cells were washed with PBS, trypsinized, harvested, and fixed with 1 mL of icecold 70% ethanol. The ethanol was discarded by centrifugation and cells were washed with PBS. The cell pellets were stained with the PI solution and RNaseA and incubated for 30 min at room temperature. The stained cells in different cell cycle phases were analyzed by FACSCalibur (Becton Dickinson).

# Apoptosis detection by DAPI staining

DAPI staining is used for observing condensed and fragmented nuclei of the cells for qualitative study of apoptosis (Nam 2013; Kang 2015). For DAPI staining, MCF-7 cells were seeded in a 6-well plate ( $5 \times 10^4$  cells/well) and incubated at 37 °C in 5% CO<sub>2</sub> overnight. Cells were then treated with IC<sub>50</sub> concentration of samples (GNCs, GNCs@Dox-Pol, HGNSs, HGNSs@Dox-Pol, and free Dox). After 4 h, the medium was removed carefully and washed with PBS to eliminate un-uptaken GNSs. Culture medium was

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added again and further incubated for 24 h. Then, cells were irradiated and incubated for 3 h. Subsequently, cells were implemented to fixation by 10% paraformaldehyde. After washing with PBS, Triton X-100 (0.1% w/v) was added for 15 min to increase the permeability of the cell membrane. Identification of apoptotic morphological alteration and DNA fragmentation of cells were investigated by staining the cells nuclei with DAPI under a fluorescence microscope (Olympus microscope Bh2-RFCA, Japan).

# Apoptosis detection by flowcytometric method

To determine apoptotic cell's value, cells were seeded in six-well plates with a density of  $5\times10^5$  cells per well in the culture medium and incubated for 24 h. Then, the cells were treated with the IC50 concentration of samples (GNCs-Pol, GNCs@Dox-Pol, HGNSs-Pol, HGNSs@Dox-Pol). After 4 h, the medium was removed carefully and washed with PBS to eliminate un-uptaken GNSs. Culture medium was added again and incubated for 24 h. Then, cells were irradiated and were further incubated for 3 h. Non-treated cells were considered as control. After 24 h, the cells were trypsinized, collected, and finally, they centrifuged and dispersed in 1 ml PBS. The cells were suspended in 100  $\mu$ L annexinbinding buffer and 2.5  $\mu$ L Annexin-V and 2.5  $\mu$ L propidium iodide (PI)was added to cell suspension for 14 min and kept in dark at room temperature (according to ex-bio kit manufacture protocol). Finally, a FACS Calibur flow cytometer was used for the percentage of early and late apoptosis Annexin-V/PI apoptosis assay.

# **Results and discussion**

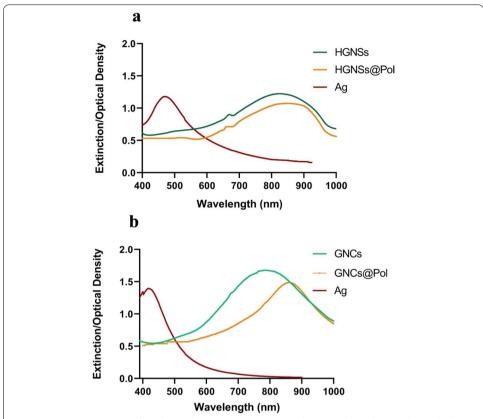
# Characterization of gold nanostructures and micelles

The UV–Vis spectra verified the evolution process of both GNSs (Fig. 1). By optimizing the experimental conditions, the SPR peak of HGNSs and GNCs were tuned in the NIR region (Fig. 1a). UV–Vis spectra of Ag colloid that was used as seed nanoparticles for the synthesis of HGNSs showed an SPR peak in 468 nm. This peak shifted to the red region (820 nm) by adding AuCl4<sup>-</sup> (the sign of HGNSs formation). Coating of HGNSs with polymer shifted the peak to 848 nm (owing to light scattering of a polymeric shell). UV–Vis spectra of Ag nanocubes showed an SPR peak in 416 nm. For the synthesis of GNCs, galvanic replacement of Au/Ag ions was carried out which lead to a redshift in SPR peak (782 nm). Similarly, coating of GNCs with polymer also shifted the UV–Vis spectra peak to 856 nm (Fig. 1b).

TEM images of free micelles and micelle-coated GNSs are shown in Fig. 2. The average particle size of free micelles was around 50 nm. For GNCs, and HGNSs, the size in wall length were 40-50 and 20-25 nm, respectively (interior pore size for GNCs and HGNSs were 2-4 nm and 1-2 nm, respectively). After encapsulation by copolymer, a light shell appeared around the GNSs as can be seen in Fig. 2d, g.

Particle size, surface charge, and polydispersity index (PDI) of the GNSs and GNSs@ Pol are shown in Table 1 and Fig. 3. The surface charge of GNSs was different in each stage which was verified by zeta potential ( $\zeta$ ). Since the surface of GNCs and HGNSs were covered with PVP and citrate,  $\zeta$  potential of GNCs and HGNSs were neutral and negative, respectively (1.4 and - 10 mV). Coating of GNCs and HGNSs with polymer caused a decrease in zeta potential to - 28.8 and - 24 mV, respectively. For nanoparticles, zeta potential value could determine their stability (higher positive or negative zeta

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**Fig. 1** UV–Vis spectroscopy of synthesized GNSs. **a** UV–Vis spectra of HGNSs with/without polymer shell. In UV–Vis spectra, maximum absorption of HGNSs and HGNSs@-Pol were 820 and 848 nm, respectively, and polymer-coated HGNSs showed a redshift. **b** UV–Vis spectra of GNCs with/without polymer coating. The maximum absorption of GNCs was about 782 nm and for polymer-coated GNCs was 856 nm that proved a redshift

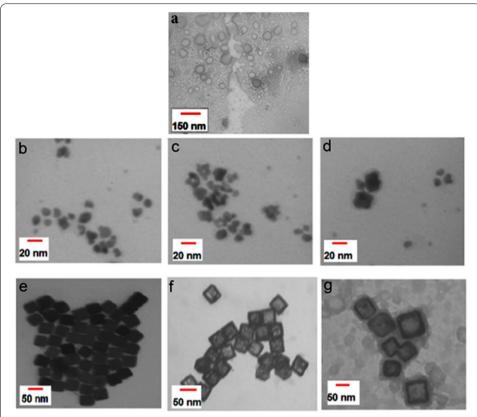
potential indicates higher stability) (Honary and Zahir 2013; Jalilzadeh 2020). PDI confirmed the narrow distribution of the GNSs and GNSs@Pol.

Figure 4a shows the stability of GNSs and GNSs@Pol in serum-containing buffer solution after 2 months of keeping at 4 °C. The solution of HGNSs@Pol was found to be monodisperse after 2 months without any visible aggregation proved by particle size (128.8 nm) and PDI (0.139) captured by DLS (Table 2). Comparison of SPR peak of HGNSs@Pol right after synthesis and after 2 months of maintenance at 4 °C showed no observable change which approved its stability. However, GNCs@Pol particle size showed 65 nm increase after 2 months (Table 2). Moreover, the SPR peak of GNCs@Pol became wider after 2 months which shows a lesser stability for GNCs@Pol compared to HGNSs@Pol.

# In vitro photothermal conversion comparison

Previous studies discovered that due to the excellent tissue penetration capability of NIR laser, 808 nm laser was selected for the photothermal study. To receive enough energy from every particle inside the cells by 808 nm laser, the exposure time was increased (10 min) (Gharatape 2016; Seo 2018; Cabral and Baptista 2013). GNCs and HGNSs show special NIR optical property and can be used as photothermal agents

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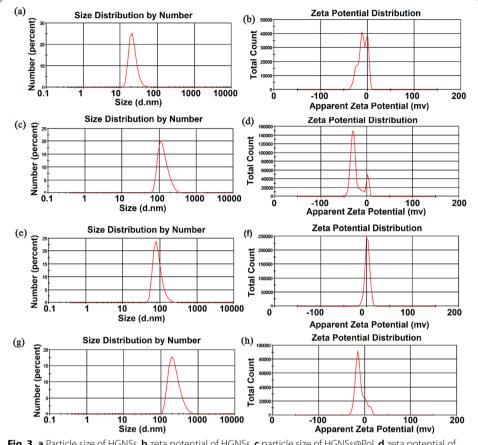
**Fig. 2** TEM image of synthesized GNSs. **a** Free micelles with an average size of around 50 nm. **b–d** Ag nanostars, HGNSs and HGNSs@Pol. **e–g** Ag nanocubes, GNCs and GNCs@Pol, respectively

**Table 1** Size, PDI, and zeta potential of the synthesized nanostructures

GNSs	Size (nm)	PDI	Zeta potential (mV)
HGNSs@Pol	132	0.478	<b>-</b> 24
GNCs	88.31	0.139	1.4
GNCs@Pol	238.8	0.344	<b>–</b> 28.5

(Wang 2012; Khan 2012). In our experiments, their photothermal conversion property was investigated via laser irradiation at  $\lambda = 808$  nm at a power density of 0.7 W cm<sup>-2</sup> in various exposure times with 100 µg mL<sup>-1</sup> concentration of GNSs. Temperature increase ( $\Delta T$ ) of GNCs and HGNSs suspension after 1 min laser irradiation was 12 and 13 °C, respectively (Fig. 5). As expected, after 10 min, temperature rise ( $\Delta T$ ) for GNCs showed 25 increase and for HGNSs showed 27.5 °C increase, whereas the temperature of pure water only showed 3 °C increase under similar conditions. The obtained results showed that after exposure of GNSs solution to laser irradiation for 10 min,  $\Delta T$  of HGNSs was higher than GNCs. Hence, both these structures have the potential for photothermal conversion, and HGNSs proved to have a slightly higher

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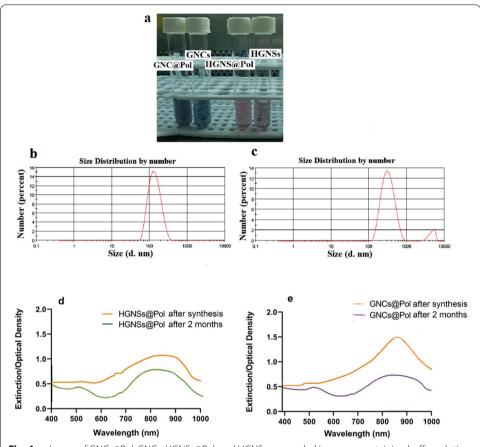
**Fig. 3** a Particle size of HGNSs, **b** zeta potential of HGNSs, **c** particle size of HGNSs@Pol, **d** zeta potential of HGNSs@Pol, **e** particle size of GNCs, **f** zeta potential of GNCs, **g** particle size of GNCs@Pol and **h** zeta potential of GNCs@Pol obtained by DLS

potential. This is likely due to the presence of multiple sharp spikes in HGNSs, which act like lightning rods that dramatically enhance the local electromagnetic field (Liu 2015). Therefore, we observe that the photothermal conversion efficiency depends on the shape and morphology of the gold nanoparticles. This is in accordance with the Wang et al. study which reported a higher temperature boost in gold nanocages compared to gold nanospheres (Wang 2015). Likewise, Moustaoui et al. compared the temperature rise of spherical and urchin-shaped gold nanoparticles and demonstrated that branched nanoparticles display a higher temperature elevation than spherical shaped nanoparticles Moustaoui 2019).

# **Biocompatibility of polymer-coated GNSs**

Hemolysis assay was applied to determine the biocompatibility of nanoparticles. Minimal hemolysis after nanoparticles interaction with red blood cells (RBCs) is a sign of their biocompatibility (Rahimi et al. 2017). Since HGNSs have better PTT ability, they were chosen for hemolysis assay to estimate the blood adaptability of HGNSs-coated ABC triblock copolymer (HGNSs@Pol) on RBCs (Fig. 6a,b). In this concern, RBCs were treated with various concentrations of HGNSs@Pol (1.5–100  $\mu$ g mL $^{-1}$ ) as well as PBS buffer (pH 7.4), and deionized water as the negative and positive control,

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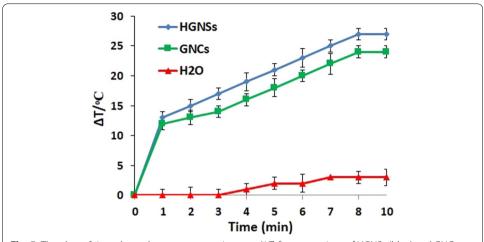
**Fig. 4** a Image of GNCs@Pol, GNCs, HGNSs@Pol, and HGNSs suspended in serum-containing buffer solution after 2 months at 4 °C (pH 7.4); particle size of **b** HGNSs@Pol and **c** GNCs@Pol after 2 months; Comparison of SPR peak right after synthesis and after 2 months of keeping in 4 °C for **d** HGNSs@Pol and **e** GNCs@Pol.

**Table 2** Particle size and PDI of the HGNSs@Pol and GNCs@Pol after 2 months

GNSs	Size (nm)	PDI
HGNSs@Pol	128.8	0.139
GNCs@Pol	307.2	0.344

respectively. Treatment of RBCs with PBS buffer did not modify the normal cells form, but deionized water-induced complete hemolysis of erythrocytes. The absorption of supernatant samples was measured at 540 nm (hemoglobin) by UV–Vis. The results showed that in physiological conditions, a high concentration of HGNSs@Pol (100  $\mu g$  mL<sup>-1</sup>) did not have any hemolytic effect. But in 200  $\mu g$  mL<sup>-1</sup> which is the IC50 concentration of free micelles, a slight hemolytic effect was observed (Fig. 6a). Optical microscope imaging showed that RBCs shape do not have significant changes after treatment for 3 h and their disc-like structure has been preserved (Fig. 6b). Therefore, it can be concluded that HGNSs@Pol are suitable in drug delivery systems and other in vivo applications.

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# **Fig. 5** The plots of time-dependent temperature increase ( $\Delta T$ ) for suspensions of HGNSs (blue) and GNCs (red) as a function of irradiation time using a 808 nm laser with a power density of 0.7 W cm<sup>-2</sup>

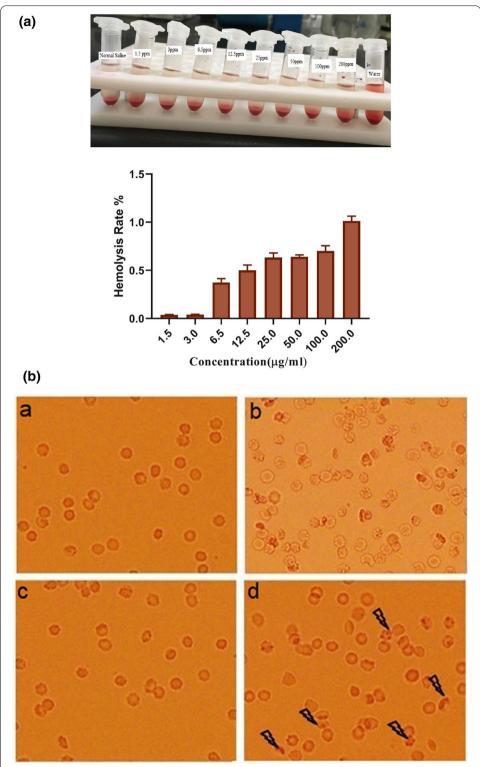
# Loading and in vitro release of Dox

The influences of pH and laser irradiation in the Dox release profile of GNSs@Pol were measured by UV–Vis spectroscopy similar to the method used in Liao (2015). According to the standard absorption curve of Dox ( $\lambda$  max=480 nm), entrapped efficiency of Dox in HGNSs@Pol and GNCs@Pol were 78.02% and 84.3%, respectively.

# pH-responsive release of Dox from GNSs@Dox-Pol

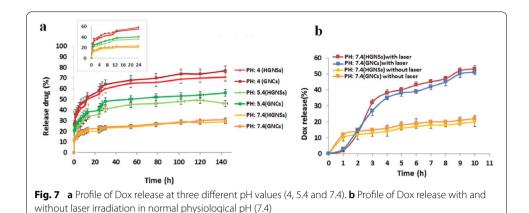
In order to investigate pH-responsive release profile, Dox release from HGNSs@Pol and GNCs@Pol was investigated at three different pH values (4, 5.4 and 7.4). At pH 4, over 58 and 55% of Dox were released from the GNCs@Pol and HGNSs@Pol in 24 h, respectively. As shown in Fig. 7a, the drug released amounts in the pH 5.4 and 7.4 were 36 and 20% for HGNSs@Pol and 40 and 23% for GNCs@Pol in 24 h, respectively. After 6 days of incubation at pH 4, 77 and 71% of Dox was released from GNCs@Pol and HGNSs@ Pol, respectively. At the same period of time at pH 5.4, 56 and 46% of Dox was released from GNCs@Pol and HGNSs@Pol, respectively. At physiological pH (7.4) after 6 days only 30% of Dox was released from both formulations. Results showed that at all values of pH, Dox release values were higher for GNCs@Pol than for HGNSs@Pol. This could be explained by larger pore diameters and a higher number of pores in GNCs which makes the release of Dox easier (Sun 2017). pH-dependent Dox release behavior in both nano-formulations can be explained by the formation and elimination of ionic interaction between Dox and nanocarrier by variation of release medium pH. Under physiological conditions (pH: 7.4), Dox release is limited due to the presence of ionic interaction between the protonated amine group of Dox (pka=8.3) and de-protonated carboxylic acid groups of PAA (pka=5.6). Whereas at pH  $\leq$  5.4 carboxylic acid groups of PAA are protonated (Pka = 5.6) and ionic interaction between nanocarrier and Dox was removed, and hence Dox was released (Salehi et al. 2015). Abbasian et al. showed that Dox release from GNRs@Dox-Pol at pH 4, 5.4, and 7.4 after 150 h (at 37 °C) was 70, 50.8, and 22%, respectively. They also showed that by increasing temperature to 45 °C Dox release

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**Fig. 6 a** Hemolysis rate of HRBCs in different concentrations of ABC triblock copolymer (1.5, 3, 6.5, 12.5, 25, 50, 100 and 200 μg mL<sup>-1</sup>). **b** Representative images of the hemolytic test under light microscopy: **a** normal saline, **b** distilled water, **c**100 μg mL<sup>-1</sup>, and **d** 200 μg mL<sup>-1</sup> of HGNSs@Pol

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increased to 85, 60, and 31.65%, respectively, at the same pH values and period of time (Abbasian 2017).

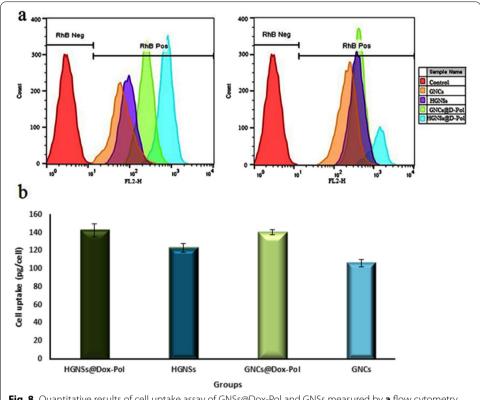
## Effect of laser irradiation on the Dox release from GNSs@ Dox-Pol

Dox release from GNSs@Dox-Pol was significantly accelerated by laser irradiation that can be explained by temperature-sensitive properties of the copolymer (Xia 2011). The lower critical solution temperature (LCST) of the ABC copolymer was 41 °C (Abbasian 2017). In fact, at temperatures higher than LCST, PNIPAAm chains collapse, and hence opens the pores and leads to a release of the preloaded Dox. The heat needed for this process was provided by laser irradiation (Cobley 2010). As shown in Fig. 7b, over 55% of drug released happens in the first 10 h (released amounts would be increased in HGNSs due to better light to heat conversion). When the laser is turned off the polymer chains will relax back to the extended state, and thereby interrupting the release. Our results were in accordance with the study by Yavuz et al. that the release rate of Dox from GNCs coated with thermo-responsive polymers was accelerated by laser irradiation (Yavuz 2009; Li 2011).

#### Intracellular uptake in vitro

The physicochemical properties of nanostructures could influence cellular uptake (Jiang 2015; Lara-Cruz 2019; Dykman and Khlebtsov 2014). In PTT, the higher cell uptake of GNSs (as heat transducers), the higher chance of transduction of photon energy into heat. In vitro uptake of GNSs into MCF-7 cells was investigated by flow cytometry, fluorescent microscopy and ICP-MS. Flow cytometry provides a quantitative measure of cellular uptake of GNs@Dox-Pol and GNSs. The fluorescent intensity of rhodamine B corresponds to the internalized amounts of GNSs@Dox-Pol and GNSs. Although GNSs (GNCs and HGNSs) treatment group showed slight fluorescence intensity, coating with Dox-Pol (GNSs@Dox-Pol) significantly increased this intensity. This indicates that the copolymer coating, as the biocompatible outer layer, had a significant positive effect on cellular uptake of GNSs (Fig. 8a). This confirms the report by Sun et al. that the 4T1 cell membranes coated on the Dox-loaded gold nanocages can increase the uptake of cancer cells by enhancing the fluorescence intensity (Sun 2017).

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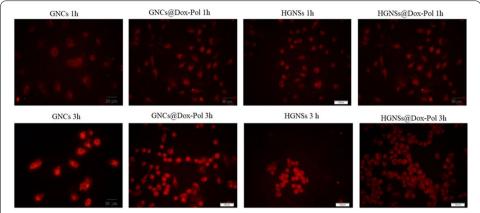


**Fig. 8** Quantitative results of cell uptake assay of GNSs@Dox-Pol and GNSs measured by **a** flow cytometry and **b** by ICP-MS

Using ICP-MS, the intracellular Au content of samples was measured quantitatively. As shown in Fig. 8b, the cell uptake of HGNSs after 3 h of incubation, was higher than that of GNCs (124 and 107 pg/cell, respectively) due to the small size of HGNSs compared to GNCs (refer to Table 1). Another reason is that the branched shape of HGNSs has a higher probability to enter the cell in comparison with the cube-like morphology of GNCs which is consistent with previous studies (Wang 2013). After 3 h of incubation, the cell uptake of HGNSs@Dox-Pol and GNCs@Dox-Pol was 142.85 and 140.84 pg/cell, respectively. These results indicated that the polymer coating on GNSs can increase the potential of cell internalization, maybe due to the biocompatible nature of copolymer. This result is in accordance with the Wang et al. study, which compared the cellular uptake of different shapes of Au nanostructures (nanorod, nanocage and nanohexapod) using the ICP-MS method. Their results revealed the impact of gold nanostructure morphology on their uptake by cells. Their results showed that the cellular uptake of Au nanohexapods was significantly higher than other morphologies of gold nanostructures (Wang 2013).

To further confirm cellular uptake of the nanostructures, qualitative uptake of GNSs and GNSs@Dox-Pol was monitored by fluorescence microscopy. The cells were treated with rhodamine B-loaded GNSs (GNCs and HGNSs) and GNSs@Dox-Pol (GNCs@Dox-Pol and HGNSs@Dox-Pol) for 1 and 3 h. Figure 9 shows that rhodamine B intensity was increased from 1 to 3 h. This result demonstrated the time-dependent gathering of the nanostructures into the cells. More fluorescent intensity (cell uptake) was observed for

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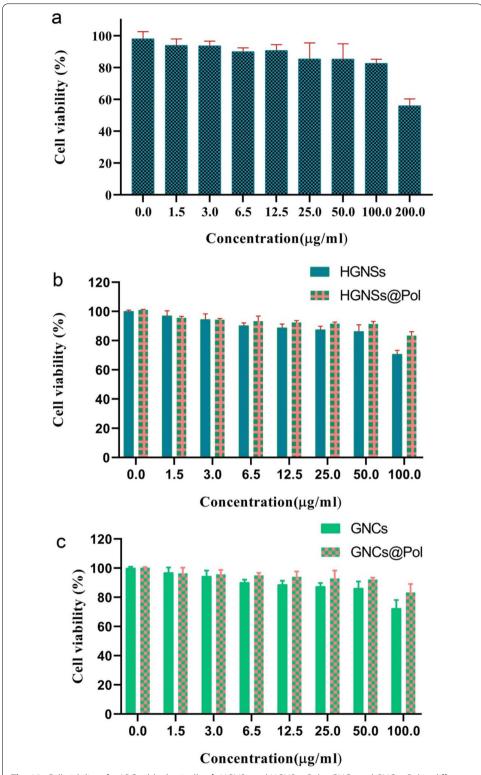
**Fig. 9** Fluorescent microscopy images of Rhodamine B-labeled GNSs and GNSs@Dox-Pol uptake after 1 and 3 h of incubation in MCF-7 cancer cells

HGNSs and HGNSs@Dox-Pol in comparison with GNCs and GNCs@Dox-Pol, respectively, which can be explained by size-dependent cell uptake (Sonavane et al. 2008). Similar to results from flow cytometry and ICP-MS (Fig. 8), fluorescent microscopy results showed high fluorescent intensity for HGNSs@Dox-Pol and GNCs@Dox-Pol compared to HGNSs and GNCs. This can be explained by the biocompatibility of the copolymer. Liao et al. examined cellular uptake of the GNRs-loaded polymersomes with fluorescent microscopy and suggested that polymersomes can decrease the cytotoxicity of the GNRs and increase cell uptake efficiency (Liao 2015). The red color intensity was increased from 1 to 3 h indicating the time-dependent cell uptake of the nanoparticles into the cells. HGNSs@Dox-Pol had a stronger fluorescent intensity when compared with GNCs@Dox-Pol in cells which may be associated with small size (refer to Table 1) and branched morphology of HGNSs.

# Cytotoxicity assay

The dose-dependent cytotoxicity of the bare GNCs and HGNSs toward MCF-7 cells was first evaluated by MTT assay (Broek 2011; Liao 2015). In vitro biocompatibility and cytotoxic effects of the ABC triblock copolymer, GNSs and GNSs@Pol were evaluated by MTT assay against MCF-7 cells for concentrations ranging from 1.5 to 100  $\mu$ g mL<sup>-1</sup> at the end of 48 h of the incubation period. As shown in Fig. 10, after 48 h, the cell viability of copolymer was over 80%, even at a high concentration of 100 μg mL<sup>-1</sup>. This indicates that the synthesized copolymer was not toxic to MCF-7 cells. GNCs and HGNSs had an insignificant effect on MCF-7 cancer cells even at high concentrations (cell viability of GNSs was above 70% at 100 μg mL<sup>-1</sup>) which is in agreement with a report by Wang et al. (2014). After polymer coating (GNSs@ Pol) cell viability increased to a minimum of 90% in the same concentrations. Hence, coating of GNSs with copolymer increases its biocompatibility. Kumar Rengan et al. compared the biocompatibility of gold nanocages and hollow gold nanoshells under similar conditions (from 1.5 to 100 μg mL<sup>-1</sup>) in the L929 mouse fibroblast cell line. They found that the cell viability of the gold nanocages was above 90% at 100  $\mu$ g mL<sup>-1</sup> concentration. For gold nanoshells, they showed that, at the same concentration, the

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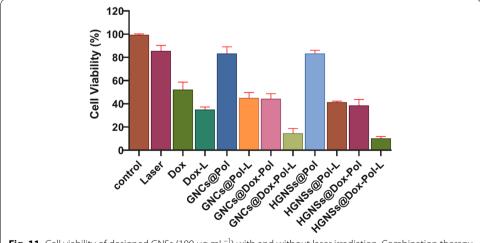
**Fig. 10** Cell viability of **a** ABC triblock micelles, **b** HGNSs and HGNSs@Pol, **c** GNCs and GNCs@Pol in different concentrations (1.5, 3, 6.5, 12.5, 25, 50, and 100  $\mu$ g mL<sup>-1</sup>) over 48 h treatment of MCF-7 cell line evaluated with MTT assay

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cell viability decreased slightly to 80% (Rengan 2014). Robinson et al. compared the effect of gold nanorods and nanocages for 48 h in DU145 cells at optical densities ranging from 0 to 12. Their results showed that cell viability in GNRs and GNCs treatment groups were not significantly different if CTAB is thoroughly removed from gold nanorods (Robinson et al. 2015).

# Cytotoxicity assay by laser irradiation

Photo/chemo-toxicity of designed GNSs was evaluated in MCF-7 cells by MTT assay. As shown in Fig. 11, GNCs and HGNSs showed slight toxicity (cell viability of GNCs and HGNSs were 72.4 and 70.65%, respectively), and polymer coating decreased the toxicity of nanostructures (cell viabilities reached ~83%). In the first monotherapy groups, that received only PTT, cell viability of GNCs@Pol and HGNSs@Pol after NIR irradiation decreased to 45.1 and 41.4%, respectively. In the second monotherapy groups, that received only chemotherapy, cell viability of Dox-loaded GNCs@Pol and HGNSs@Pol without laser irradiation, decreased to ~44.4% and ~38.6%, respectively. In combination therapy groups that received chemo-photothermal therapy, cell viability dropped to ~14.6% and 10.2% for GNCs@Dox-Pol and HGNSs@Dox-Pol, respectively. This high rate of mortality of MCF-7 cells indicates a synergistic effect of PTT and chemotherapy. On the other hand, due to the thermo-sensitivity of our polymer, Dox release was also accelerated by temperature increase due to laser irradiation. It is important to note that the slightly higher mortality rate of HGNSs@Dox-Pol-L compared to GNCs@Dox-Pol-L was due to the higher cellular uptake of HGNSs (effect of size and morphology, refer to Table 1 and Figs. 2, 3) and higher light to heat conversion efficacy of HGNSs compared to GNCs (Fig. 5). The results obtained by Wang et al. showed that Dox-loaded GNCs@hyaluronic acid upon NIR irradiation, reduced cell viability to 18% (Wang 2014). Note that the efficacy of hyperthermiatriggered release of Dox under the NIR irradiation indicates an excellent synergistic



**Fig. 11** Cell viability of designed GNSs (100  $\mu$ g mL<sup>-1</sup>) with and without laser irradiation. Combination therapy groups (HGNSs@Dox-Pol, GNCs@Dox-Pol, with laser irradiation) have the least cell viability in comparison with other groups

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effect in growth inhibition of MCF-7 cells. These results are in accordance with several recent studies (Xia 2011; Rengan 2014).

# Induction of apoptosis

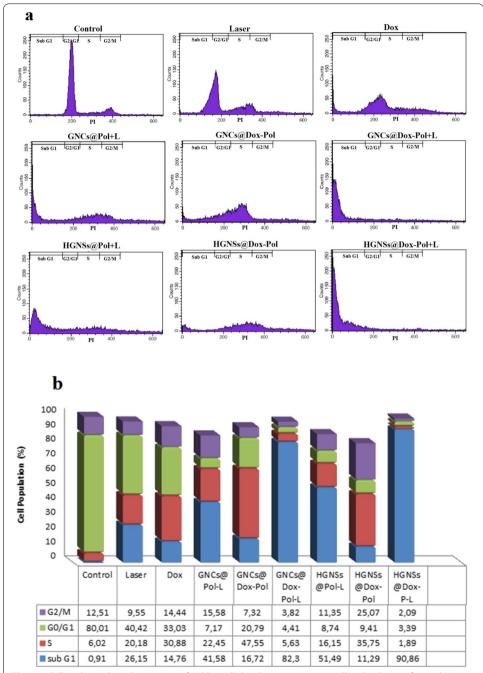
# Cell cycle arrest

Nanoparticles that are uptaken by cells can interact with cell membranes and dysregulate the cell cycle, and finally induce apoptosis or necrosis (Jones and Grainger 2009). In comparison with monotherapy, the combined PTT and chemotherapy have been reported to synergistically improve therapeutic effects in cancers and reduce side effects by decreasing the dose of toxic chemotherapeutic drugs (Chen 2013). Indeed, if a treatment can stop the cell cycle, it would be effective in cancer therapy (Ling 1996). The Sub-G0/G1 area is of special interest in many toxicity studies that show the characteristics of apoptosis (Mahmoudi 2011). In this study, we investigated the effects of administration of GNSs@Dox-Pol with/without irradiation on the cell cycle (Fig. 12a). When comparing the cell cycle pattern of non-treated MCF-7 cells (negative control group) compared to those treated with only laser irradiation (positive control group), an arrest was observed in SubG1 and to a lesser extent in S phases. In free Dox, Dox-loaded GNCs@Pol and Dox-loaded HGNSs@Pol treatment group arrest was detected in S and SubG1 phases which proves release of Dox from GNSs@Pol structures. Our results are in line with the previous reports that Dox causes S arrests (Ling 1996; Potter 2002). With laser irradiation to MCF-7 cells treated by GNCs@Pol and HGNSs@Pol, in comparison to the only laser treatment group, the percentage of arrest in SubG1 was increased from 26% to ~42 and 52%, respectively, which is a sign of cell apoptosis. The increase in the population of cells arrested in Sub G1 proves that GNSs were accumulated in cells that can successfully improve the light to heat energy conversion. Therefore, these GNSs act as transducers in PTT treatment. In cells that received both chemotherapy and PTT in GNSs@ Dox-Pol treatment groups, the percentage of sub G1 cell population was increased drastically. A synergistic effect was observed after the treatment of cells with GNSs@Dox-Pol with laser irradiation. 90.86% and 82.30% of cells were shifted to the SubG1 phase after treatment with HGNSs@Dox-Pol+L and GNCs@Dox-Pol+L, respectively.

# DAPI staining assay

Nuclear staining using 4,6-diamidino-2-phenylindole (DAPI) is the well-known standard test for the evaluation of healthy, apoptotic, and necrotic cells through assessment of chromatin morphology and its density (Jafarirad 2018). The cell groups treated with GNSs@Pol in absence of laser exposure and the cells treated with irradiation alone (as positive control) would not induce significant chromatin degradation or cell death. Moreover, free Dox (with the same concentration of loaded groups) and GNSs@Dox-Pol groups did not cause large-scale chromatin degradation, whereas the MCF-7 cells that received combination therapy (GNSs@Dox-Pol and laser) showed tremendous chromatin condensation and fragmentation along with a significant decrease in the cell population. The synergistic effect of combination therapy was more dominant in HGNSs@Dox-Pol+laser compared to GNCs@Dox-Pol+laser. Poursalehi et al. showed that monotherapy on MCF-7 cells (treating with free Dox or Dox-loaded gold nanocomposites without laser exposure) were not able to induce

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**Fig. 12** Cell cycle results. **a** Percentage of MCF-7 cell distribution in various cell cycle phases after 48 h treatment with Dox, GNCs@Pol+laser, GNCs@Dox-Pol, GNCs@Dox-Pol+laser, HGNSs@Pol+laser HGNSs@Dox-Pol, HGNSs@Dox-Pol+laser. Cells treated with laser only (808 nm, 0.7 W.cm<sup>-2</sup>, 10 min) were considered as positive control and non-treated cells were considered as negative control. **b** Quantitative results of cell cycle arrest

apoptotic cell death (changes in chromatin morphology and fragmented nucleus) as much as combination therapy (Poursalehi 2019). In another study by Agabeigi et al., MCF-7 cells treated with MTX and FA-loaded Au@SiO<sub>2</sub> NPs that were exposed to low-level laser irradiations (combination therapy group) induced significantly more

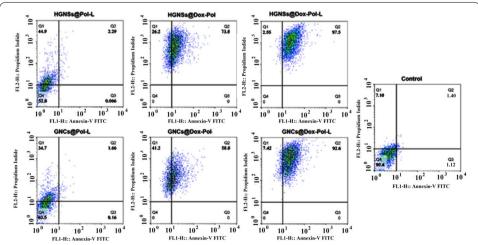
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apoptotic cells, compared to the cells groups that received monotherapy (chemotherapy or PTT alone) (Agabeigi 2020). Another study carried out by Jafarirad et al. showed that MCF-7 cells treated with functionalized Ag/ZnO and Nd/ZnO reduced graphene oxide nanocomposites which were exposed to low-level laser irradiation were able to induce apoptosis more effectively than cells treated with nanocomposites or laser alone (Jafarirad 2018).

# Apoptosis analysis by the flow cytometry method

In order to have a quantitative evaluation on the effect of GNSs morphology in PTT efficacy, and the combination effect of chemo-PTT on the induction of apoptosis in MCF7 cells, Annexin-V apoptosis assay was done. The greater influence of DNA damage has been ascribed to a stop the replication fork, leading to double-strand breaks production, which reduces cell viability (Agabeigi et al. 2020).

Figure 13 presents the results of the Annexin V-FITC assay. MCF-7 cells treated with HGNSs@Pol and GNCs@Pol with laser irradiation (monotherapy: photothermal treatment group) showed 45 and 35% of necrotic cells and the viable cells population fall to 53 and 63%, respectively. The presence of gold nanoparticles as photo-synthesizer is essential in photothermal therapy and HGNSs because of higher cellular uptake and heat generation ability showed more destructive effect than GNCs. In the cells treatment group that received HGNSs@Dox-Pol and GNCs@Dox-Pol without laser irradiation (monotherapy: chemotherapy treatment group), 74 and 59% of cells showed apoptosis and 26 and 41% necrosis, respectively. The results indicated that HGNSs@Dox-Pol was more successful in the induction of apoptosis (programmed cell death) compared to GNCs@Dox-Pol may be due to the higher Dox release in cells because of higher cellular uptake of HGNSs@Pol in comparison with GNCs@Pol. MCF-7 cells treated with HGNSs@Dox-Pol-L and GNCs@Dox-Pol-L that received combination therapy (chemo-photothermal therapy) showed the highest apoptosis and lowest necrotic cell death (97.5 and 92.6% apoptosis and 2.5 and 7.42% necrosis,



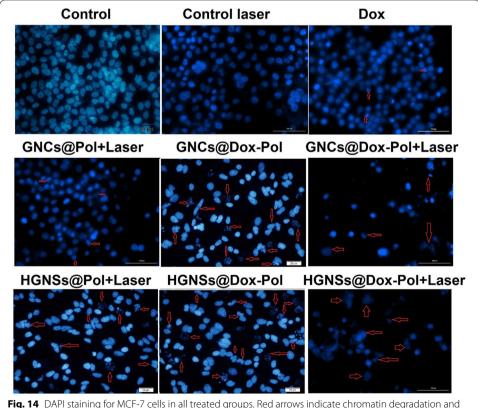
**Fig. 13** The apoptotic effects of MCF7 cells, obtained by flow cytometry, for cells treated with different monotherapy treatment groups (chemotherapy or photothermal therapy) or combination therapy groups. Cells with no treatment were considered as control. L: laser irradiation (808 nm. 07 W/cm²)

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respectively). The maximum apoptosis rate (97%) was detected in cells treated with HGNSs@Dox-Pol. As a result, the photothermal efficacy of these GNSs was dependent on the morphology of the nanostructure. The results of the apoptosis assay confirmed the results obtained in cell cycle assay (Fig. 12), DAPI staining (Fig. 14) and cell viability assay after laser irradiation (Fig. 11) that all were related to the results obtained in cell uptake and temperature elevation studies (Figs. 8 and 5). The more particles absorbed into the cell, the greater the number of nanoparticles inside the cell that convert the photon energy into heat. This heat generated inside the cell destroys the cell. Also, the efficiency of converting photon energy to heat is another important factor that is completely dependent on the morphology of gold nanoparticles. The higher ability to convert photon energy to heat in a structure, the higher the photothermal efficiency of that structure and the more toxicity is observed (due to the higher intracellular heat production). In this study, both anisotropic gold nanostructures (HGNSs and GNCs) showed high toxicity due to their structural potential for high intracellular heat production and it was slightly better in HGNSs.

#### Conclusion

In this work, we investigated combination chemo/PTT therapy with GNCs and HGNSs-coated thermo/pH-sensitive polymers. GNCs and HGNSs were selected for their efficient light to heat transduction and extensive Dox delivering capacity upon



**Fig. 14** DAPI staining for MCF-7 cells in all treated groups. Red arrows indicate chromatin degradation and fragmentation

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laser irradiation. HGNSs and GNCs were synthesized, coated with PAA-b-PNIPAAm-b-PCL-SH, and loaded with Dox. GNSs@Pol has a high capacity to accumulate in MCF-7 cells that were proved by different cell uptake assays. Dox release accelerated in tumor cells acidic environment and under NIR laser irradiation. GNSs@Dox-Pol upon laser irradiation (combination therapy group) increased tunable therapeutic efficacy compared to monotherapy (chemotherapy or PTT alone). The increased population of apoptotic cells was confirmed by DAPI staining, cell cycle studies and Annexin V–PI assays. In combination therapy with both gold nanostructures, hallow gold nanostars showed superior efficacy compared to gold nanocages. Three major reasons may be suggested for this superiority. First, HGNSs had a smaller size than GNCs that raised cellular uptake of HGNSs compared to GNCs. Second, spindle-like morphology of HGNs can lead to superior cellular uptake compared to cube shape GNCs. Accumulation of GNSs in cells has a straight impact on PTT efficacy. Finally, HGNSs has often been advocated as more efficient light absorbers than GNCs due to their branched structure that tuned plasmon resonances.

#### **Abbreviations**

PTT: Photothermal therapy; GNSs: Gold nanostructures; HGNSs: Hollow gold nanostars; GNCs: Gold nanocages; DLS: Dynamic light scattering; TEM: Transmission electron microscopy; Dox: Doxorubicin; PDI: Polydispersity index; LCST: Lower critical solution temperature; MTT: 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide; DAPI: 4,6-Diamidino-2-phenylindole.

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#### Authors' contributions

AP: gold nanoparticles synthesis and data collection and figure preparation. MA: conducting in data analysis. FR: contributed to the edition of manuscript. FB: contributed in laser experiments. FM: synthesis of the copolymer. RS: proposal design and ideas, contributed to the cellular tests, data analysis and writing of manuscript. MM: contributed in the organization and writing. All authors read and approved the final manuscript.

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## Availability of data and materials

Not applicable.

#### **Declarations**

#### Ethics approval and consent to participate

Not applicable.

#### Consent for publication

Not applicable.

# **Competing interests**

The authors declare that they have no competing interests.

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